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Metal halide lamp

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Metal halide lamp

The invention relates to an automotive metal halide lamp comprising a substantially cylindrical discharge vessel having an internal diameter $D_i < 2.0$ mm, and filled with an ionizable filling, wherein two electrodes are present having a mutual distance E_A , wherein preferably $3 \text{ mm} < E_A < 7 \text{ mm}$, for maintaining a discharge in the discharge vessel, and wherein the filling comprises an inert gas such as Xe having a pressure at room temperature between 5 and 25 bar, and an ionizable salt.

Such a lamp is described in the international patent application WO 00/67294.

~~10 Many modern automotive metal halide lamps have a very small, very high pressure discharge~~
vessel surrounded by a gas filled outer bulb, and having a lamp power between 20 W and 40 W. The filling of the lamp can contain Hg, or alternatively can be mercury free and contain Zn or ZnI_2 . Such lamps require highly efficient ionizable salts, and it is known to use a salt mixture of NaI and CeI_3 . Such a lamp is based on the recognition that a high efficacy and a
15 sufficiently high color rendering is possible when sodium halide is used as a filling ingredient of a lamp and a strong widening and inversion of the Na emission in the Na-D lines takes place during lamp operation. This requires a high coldest-spot temperature in the discharge vessel, which excludes under practical conditions the use of quartz or quartz glass for the discharge vessel wall and renders the use of a ceramic material for the discharge vessel wall
20 preferable. The term "ceramic wall" in the present description and claims is understood to cover a wall of metal oxide such as, for example, sapphire or densely sintered polycrystalline Al_2O_3 , as well as metal nitride, for example AlN. The known lamp combines a good color rendering with a comparatively wide range of the color temperature.

The lamp has the advantage that the discharge vessel has very compact
25 dimensions which render the lamp highly suitable for use in a headlamp for a motor vehicle. Owing to the small internal diameter in comparison with the electrode spacing, and thus the discharge arc length, the discharge arc is hemmed in by the discharge vessel wall, so that the discharge arc has a sufficiently straight shape for it to be suitable for use as a light source for a motor vehicle headlamp. An small internal diameter D_i is found to be of essential

importance for realizing a sharp beam delineation necessary for use in motor vehicles in combination with a small spot of high brightness immediately adjacent this delineation. Such very small internal diameter renders the lamp particularly suitable for use as a light source in a complex-shape headlamp. An advantage of such a headlamp is that no separate passing-
5 beam cap is required in the formation of the light beam to be generated in order to realize a sufficiently sharp beam delineation.

The drawbacks of the known lamp are however a relatively low correlated color temperature (CCT) (between 3000 and 3500 K), a relatively unstable luminous flux, a relatively unstable wall temperature, a relatively large initial color point spread and a
10 relatively large color point shift during life time, mainly due to chemical transport and segregation of the NaI/CeI₃ salt mix.

The object of the invention is an automotive metal halide lamp wherein one or
15 ~~more of the mentioned drawbacks are alleviated. In order to achieve that goal, said ionizable~~
salt is selected from the group comprising PrI₃, NdI₃ and LuI₃. Preferably said ionizable salt further comprises NaI, wherein the molar ratio NaI/(PrI₃ + NdI₃ + LuI₃) lies between 1.0 and 10.3. Although usually one of the mentioned rare earth iodides will be used, it is possible to use a mixture as well. It was found that in a lamp of the mentioned properties these salts are
20 only slightly sensitive for big variations in lamp power and thus in coldest spot temperature, while showing a color spot close to the BBL ("black body line"), and that these salts are relatively insensitive for color shifts due to segregation, i.e. changes in the salt mix ratio at the coldest spot position of the lamp due to for instance corrosion or transport of the liquid salt. In particular the use of PrI₃ results in an excellent color temperature for automotive
25 purposes, close to the preferred CCT of 4200 K, while in the case LuI₃ is used for instance the color temperature can be enhanced by adding small amounts of Tbl₃ and/or GdI₃.

In a first preferred embodiment the molar ratio NaI/PrI₃ lies between 2.3 and 10.3, preferably between 3.0 and 5.7, more preferably is approximately 3.5. Preferably the amount of PrI₃ in the discharge vessel is between 10 and 335 $\mu\text{mol}/\text{cm}^3$, more preferably
30 between 25 and 160 $\mu\text{mol}/\text{cm}^3$, still more preferably approximately 50 $\mu\text{mol}/\text{cm}^3$. In a discharge vessel of 1.6 mm \times 3 mm (Di \times EA) this results in a CCT of approximately 4200 K. In a discharge vessel of 1.2 mm \times 6 mm the preferred concentration is 1.8 times higher in

In a second preferred embodiment the molar ratio NaI/NdI_3 lies between 3 and 6.7, preferably between 3.6 and 4.8, more preferably is approximately 4.2. Preferably the amount of NdI_3 in the discharge vessel is between 8 and $301 \mu\text{mol}/\text{cm}^3$, more preferably between 30 and $167 \mu\text{mol}/\text{cm}^3$, still more preferably approximately $45 \mu\text{mol}/\text{cm}^3$. In a discharge vessel of 1.6 mm x 8 mm (Di x EA) this results in a CCT of approximately 4200 K. In a discharge vessel of 1.2 mm x 6 mm the preferred concentration is 1.8 times higher in order to have the same CCT.

In a third preferred embodiment the molar ratio NaI/LuI_3 lies between 1.0 and 3.2, preferably between 1.2 and 1.8, more preferably is approximately 1.4. Preferably the amount of LuI_3 in the discharge vessel is between 15 and $414 \mu\text{mol}/\text{cm}^3$, more preferably between 27 and $230 \mu\text{mol}/\text{cm}^3$, still more preferably approximately $69 \mu\text{mol}/\text{cm}^3$. In a discharge vessel of 1.6 mm x 8 mm (Di x EA) this results in a CCT of approximately 4200 K. In a discharge vessel of 1.2 mm x 6 mm the preferred concentration is 1.8 times higher in order to have the same CCT.

~~The above and further aspects of the lamp according to the invention will be explained in more detail with reference to the drawings (not true to scale), wherein:~~

Fig. 1 diagrammatically shows a lamp according to the invention; and
Fig. 2 shows the discharge vessel of the lamp of Fig. 1 in detail.

Fig. 1 shows a metal halide lamp provided with a discharge vessel 3 having a ceramic wall which encloses a discharge space 11 containing an ionizable filling. Two tungsten electrodes 4, 5 whose tips 4b, 5b are at a mutual distance EA are arranged in the discharge space, and the discharge vessel has an internal diameter Di at least over the distance EA. The discharge vessel is closed at one side by means of a ceramic projecting plug 34, 35 which encloses a current lead-through conductor (Fig. 2: 40, 41, 50, 51) to an electrode 4, 5 positioned in the discharge vessel with a narrow intervening space and is connected to this conductor in a gas tight manner by means of a melting-ceramic joint (Fig. 2: 10) at an end remote from the discharge space. The discharge vessel is surrounded by an outer bulb 1 which is provided with a lamp cap 2 at one end. A discharge will extend between the electrodes 4, 5 when the lamp is operating. The electrode 4 is connected to a first electrical contact forming part of the lamp cap 2 via a current conductor 8. The electrode 5 is

connected to a second electrical contact forming part of the lamp cap 2 via a current conductor 9. The discharge vessel, shown in more detail in Fig. 2 (not true to scale), has a ceramic wall and is formed from a cylindrical part with an internal diameter D_i which is bounded at either end by a respective ceramic projecting plug 34, 35 which is fastened in a gas tight manner in the cylindrical part by means of a sintered joint S. The ceramic projecting plugs 34, 35 each narrowly enclose a current lead-through conductor 40, 41, 50, 51 of a relevant electrode 4, 5 having a tip 4b, 5b. The current lead-through conductor is connected to the ceramic projecting plug 34, 35 in a gas tight manner by means of a melting-ceramic joint 10 at the side remote from the discharge space. The electrode tips 4b, 5b are arranged at a mutual distance EA. The current lead-through conductors each comprise a halide-resistant portion 41, 51, for example in the form of a $\text{Mo--Al}_2\text{O}_3$ cermet and a portion 40, 50 which is fastened to a respective end plug 34, 35 in a gas tight manner by means of the melting-ceramic joint 10. The melting-ceramic joint extends over some distance, for example approximately 1 mm, over the Mo cermet 40, 41. It is possible for the parts 41, 51 to be formed in an alternative manner instead of from a $\text{Mo--Al}^{12}\text{O}^3$ cermet. Other possible constructions are known, for example, from EP 0 587 238. A particularly suitable construction was found to be a halide-resistant coil applied around a pin of the same material. Mo is very suitable for use as a highly halide-resistant material. The parts 40, 50 are made from a metal whose coefficient of expansion corresponds very well to that of the end plugs. Nb, for example, is for this purpose a highly suitable material. The parts 40, 50 are connected to the current conductors 8, 9 in a manner not shown in any detail. The lead-through construction described renders it possible to operate the lamp in any burning position as desired. Each of the electrodes 4, 5 comprises an electrode rod 4a, 5a which is provided with a tip 4b, 5b.

In a practical realization of the lamp as represented in the drawing a number of lamps were manufactured with a rated power of 26 W each. The lamps are suitable for use as headlamps for a motor vehicle. The ionizable filling of the discharge vessel 3 of each individual lamp comprises 30 mg/cm^3 Hg and 25 mg/cm^3 iodide, comprising NaI and a rare earth iodide chosen from the group consisting of PrI_3 , NdI_3 and LuI_3 . In a mercury free embodiment the Hg may be replaced by Zn or ZnI_2 . The filling further comprises Xe with a filling pressure at room temperature of 8 bar. The distance between the electrode tips 4a, 5a is 7 mm. The internal diameter D_i is 1.4 mm so that the ratio $EA/D_i = 3.6$. The wall

In a first embodiment the rare earth iodide is PrI_3 at approximately $50 \mu\text{mol}/\text{cm}^3$, and the molar ratio NaI/PrI_3 is approximately 3.5.

In a second embodiment the rare earth iodide is NdI_3 at $45 \mu\text{mol}/\text{cm}^3$, and the molar ratio NaI/NdI_3 is approximately 4.2.

5 In a third embodiment the rare earth iodide is LuI_3 at $69 \mu\text{mol}/\text{cm}^3$, and the molar ratio NaI/LuI_3 is approximately 1.4. In order to improve the color temperature of this lamp small amounts of TbI_3 or GdI_3 were added.

The described lamps showed excellent color temperature and color stability properties compared to NaI/CeI_3 fillings, while the efficacy is only slightly lower.

CLAIMS:

1. A metal halide lamp comprising a substantially cylindrical discharge vessel (3) having an internal diameter $D_i < 2.0$ mm and filled with an ionizable filling, wherein two electrodes are present having a mutual distance EA for maintaining a discharge in the discharge vessel, wherein the filling comprises an inert gas such as Xe having a pressure at room temperature between 5 and 25 bar, and an ionizable salt, characterized in that said ionizable salt is selected from the group comprising PrI_3 , NdI_3 and LuI_3 .
2. A lamp according to claim 1, wherein said ionizable salt further comprises NaI, and wherein the molar ratio $\text{NaI}/(\text{PrI}_3 + \text{NdI}_3 + \text{LuI}_3)$ lies between 1.0 and 10.3.
3. A lamp according to claim 2, wherein the molar ratio NaI/PrI_3 lies between 2.3 and 10.3, preferably between 3.0 and 5.7, more preferably is approximately 3.5.
4. A lamp according to any of the preceding claims 1 - 3, wherein the amount of PrI_3 in the discharge vessel is between 10 and $335 \mu\text{mol}/\text{cm}^3$, preferably between 25 and $160 \mu\text{mol}/\text{cm}^3$, more preferably approximately $50 \mu\text{mol}/\text{cm}^3$.
5. A lamp according to claim 2, wherein the molar ratio NaI/NdI_3 lies between 3.0 and 6.7, preferably between 3.6 and 4.8, more preferably is approximately 4.2.
6. A lamp according to any of the preceding claims 1 - 5, wherein the amount of NdI_3 in the discharge vessel is between 8 and $301 \mu\text{mol}/\text{cm}^3$, preferably between 30 and $167 \mu\text{mol}/\text{cm}^3$, more preferably approximately $45 \mu\text{mol}/\text{cm}^3$.
7. A lamp according claim 2, wherein the molar ratio NaI/LuI_3 lies between 1.0 and 3.2, preferably between 1.2 and 1.8, more preferably is approximately 1.4.

8. A lamp according to any of the preceding claims 1 - 7, wherein the amount of LuI_3 in the discharge vessel is between 15 and $414 \mu\text{mol}/\text{cm}^3$, preferably between 27 and $230 \mu\text{mol}/\text{cm}^3$, more preferably approximately $69 \mu\text{mol}/\text{cm}^3$.

5 9. A lamp according to any of the preceding claims 1 - 9, wherein $D_i < 1.5 \text{ mm}$.

10. A lamp according to any of the preceding claims 1 - 10, wherein EA lies between 3 mm and 7 mm.

10 11. A lamp according to any of the preceding claims 1 - 11, wherein the discharge vessel has a ceramic wall.

12. A lamp according to any of the preceding claims 1 - 12, wherein the discharge vessel is surrounded by a gas filled outer bulb.

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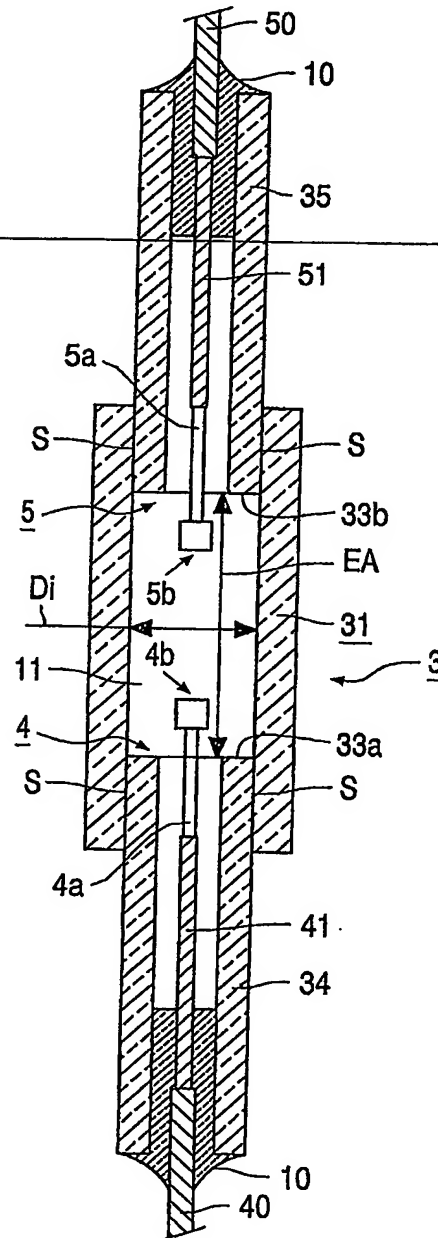
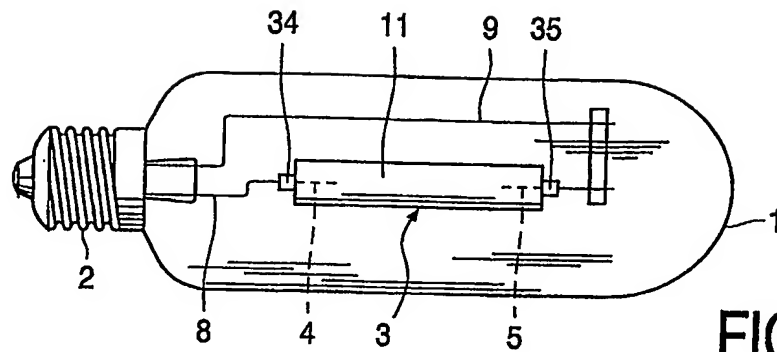
13. A lamp according to any of the preceding claims 1 - 13, wherein the lamp power lies between 20 W and 40 W.

ABSTRACT:

A metal halide lamp comprising a substantially cylindrical discharge vessel having an internal diameter $D_i < 2.0$ mm and filled with an ionizable filling, wherein two electrodes are present having a mutual distance EA for maintaining a discharge in the discharge vessel, wherein the filling comprises an inert gas, such as Xe, having a pressure at room temperature between 5 and 25 bar, and an ionizable salt, characterized in that said ionizable salt is selected from the group comprising PrI_3 , NdI_3 and LuI_3 .

Fig. 1

1/1



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